

## Chapter 34 Problem 49 †

### Given

$$v = 4.2 \times 10^5 \text{ m/s}$$

Work function of potassium  $\phi = 2.30 \text{ eV}$

### Solution

Find the wavelength of light shining on a potassium surface.

Convert the work function of potassium from electron-volts to joules.

$$\phi = 2.30 \text{ eV} \left( \frac{1.6 \times 10^{-19} \text{ J}}{1 \text{ eV}} \right) = 3.68 \times 10^{-19} \text{ J}$$

Begin with the equation for the photoelectric effect and substitute in the relationship  $c = \lambda f$  and  $K = \frac{1}{2}mv^2$

$$K_{max} = hf - \phi$$

$$\frac{1}{2}mv^2 = \frac{hc}{\lambda} - \phi$$

Solve for  $\lambda$

$$\frac{1}{2}mv^2 + \phi = \frac{hc}{\lambda}$$

$$\lambda = \frac{hc}{\frac{1}{2}mv^2 + \phi}$$

Substitute in the provided values

$$\lambda = \frac{(6.63 \times 10^{-34} \text{ J} \cdot \text{s})(3.0 \times 10^8 \text{ m/s})}{\frac{1}{2}(9.11 \times 10^{-31} \text{ kg})(4.2 \times 10^5 \text{ m/s})^2 + (3.68 \times 10^{-19} \text{ J})} = 4.43 \times 10^{-7} \text{ m} = 443 \text{ nm}$$

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†Problem from Essential University Physics, Wolfson