

## Chapter 34 Problem 43 †

### Solution

a) Find the rate of photon production for a  $1.0\text{ kW}$  radio antenna transmitting at  $89.5\text{ MHz}$ .

First find the energy per photon

$$E = hf = (6.63 \times 10^{-34}\text{ J}\cdot\text{s})(89.5 \times 10^6\text{ Hz}) = 5.93 \times 10^{-26}\text{ J}$$

Now use this value as a conversion factor between joules and photons. A watt is a joule per second. Therefore, the photon production rate is

$$P = 1.0\text{ kW} = \frac{1000\text{ J}}{\text{s}} \left( \frac{1\text{ photon}}{5.93 \times 10^{-26}\text{ J}} \right) = 1.69 \times 10^{28}\text{ photons/s}$$

b) Find the rate of photon production for a  $1.0\text{ mW}$  laser transmitting at  $633\text{ nm}$ .

First find the energy per photon

$$E = hf = \frac{hc}{\lambda} = \frac{(6.63 \times 10^{-34}\text{ J}\cdot\text{s})(3.0 \times 10^8\text{ m/s})}{633 \times 10^{-9}\text{ m}} = 3.14 \times 10^{-19}\text{ J}$$

Now use this value as a conversion factor between joules and photons. The photon production rate is

$$P = 1.0\text{ mW} = \frac{0.0010\text{ J}}{\text{s}} \left( \frac{1\text{ photon}}{3.14 \times 10^{-19}\text{ J}} \right) = 3.18 \times 10^{15}\text{ photons/s}$$

c) Find the rate of photon production for a  $2.5\text{ kW}$  X-ray machine transmitting at  $0.10\text{ nm}$ .

First find the energy per photon

$$E = hf = \frac{hc}{\lambda} = \frac{(6.63 \times 10^{-34}\text{ J}\cdot\text{s})(3.0 \times 10^8\text{ m/s})}{0.10 \times 10^{-9}\text{ m}} = 1.99 \times 10^{-15}\text{ J}$$

Now use this value as a conversion factor between joules and photons. The photon production rate is

$$P = 2.5\text{ kW} = \frac{2500\text{ J}}{\text{s}} \left( \frac{1\text{ photon}}{1.99 \times 10^{-15}\text{ J}} \right) = 1.26 \times 10^{18}\text{ photons/s}$$

---

†Problem from Essential University Physics, Wolfson