Chapter 15 Problem 18 †

Given

$$m = 8.8 \ kg$$
$$V = 0.050 \ m^3$$

Solution

a) Find the density of the compressed air.

Density is given by the formula

$$\rho = \frac{m}{V}$$

Substitute in the given values and we have

$$\rho = \frac{(8.8 \ kg)}{(0.050 \ m^3)} = 176 \ kg/m^3$$

b) What is the volume if the density is $1.2 kg/m^3$?

Use the equation given above and solve for volume

$$V = \frac{m}{\rho}$$

$$V = \frac{8.8 \ kg}{(1.2 \ kg/m^3)} = 7.33 \ m^3$$

 $^{^\}dagger \text{Problem}$ from Essential University Physics, Wolfson